Course Syllabus  
CHEM 3421 and CHEM 3021  
Quantitative Analysis and Quantitative Analysis Laboratory  
Fall 2023

There is a single grade issued for both the laboratory and lecture; hence a single syllabus.

Course Description: Analytical applications of solution chemistry.

Number of Credit Hours: 4 semester hours - 3 hours lecture and 1 hour of laboratory. A single grade is given for both laboratory and lecture.

Course Prerequisites and Corequisites: Prerequisite: CHE 1312 and 1112. Lab fee required.

Program Learning Outcomes:
1. The student will demonstrate knowledge of fundamental content in the basic areas of chemistry: Analytical, Biochemistry, Inorganic, Organic, and Physical.
2. The student will integrate knowledge with critical thinking to solve problems.

General Education Core Curriculum Objectives: There are no specific general education core curriculum objectives in this course. This course is not a general education core curriculum course.

Course Objective: To provide students with a more detailed explanation of the basic concepts, laws, and theories of some of the topics discussed in General Chemistry and to apply the knowledge to chemistry problem solving at an advanced level. The student will develop an appreciation for chemistry as it relates to the other disciplines. Furthermore, the student will recognize how chemistry provides solutions to contemporary, historical, technological, and societal issues.

Time Requirements:

CHEM 3421 is a 4-credit course and typically meets for 150 minutes a week for 15 weeks plus meets for a 2-hour final examination. The laboratory meets for 14 weeks for 3 hours. The laboratory has significant outside work in preparing and mastering the material outside of class. The course has significant weekly reading and homework assignments involving critical thinking and quantitative reasoning. Problems and homework assignments will be given. Students are assessed the material via assignments and several exams during the semester including a comprehensive final exam. For the course, these activities average at a minimum 6 hours of work each week to prepare outside of classroom hours. For the laboratory, these activities average a minimum 2 hours outside of laboratory time.
Chemistry 3421-001 and Chemistry 3021 020 and 021

Instructor’s Name:
Dr. Darrell R. Fry

Course Meeting Location:
Lehmann Chemistry Building Room 106

Course Meeting Times:
MWF 8:50 and M 4:45

Laboratory Meeting Location
Lehmann Chemistry Building Room 304

Laboratory Meeting Times:
Section 020: Thursday 2-4:50
Section 021: Friday 2-4:50

Department:
Chemistry & Biochemistry

Office:
Bush M-120

Email:
frydr@sfasu.edu

Office Hours:
MWF 9-10:30 and by appointment. Available in office or through Zoom. (Office hours are also known as student hours.)

Desire2Learn:
http://d2l.sfasu.edu
Do not email Dr. Fry through the D2L System

Homework:
https://darrellrfry.com

ZOOM LINK for Office Hours
https://sfasu.zoom.us/j/97206170593?pwd=czlJcjl5LzlhdlpTVWVxYXJSMG5hZz09

CATALOG DESCRIPTION: Quantitative Analysis – 4 semester hours, 3 hours lecture, 3 hours per week. Analytical applications of solution chemistry. Lab fee required.

PREREQUISITES: A grade of C in both CHE 133 and CHE 134. Co-requisite: CHE 3021L.

REQUIRED Texts and Other Materials:
2. The laboratory will pull heavily from chapter 2 of the required text.
   a. Scientific Calculator that is easy to do statistics with. For instance, the TI-8* graphing calculators work well for this (TI-80, TI-81, etc...).

SUPPLEMENTARY Readings:
• The Art and Science of Chemical Analysis by Christie G. Enke.
• Any freshman chemistry text such as Chemistry the Central Science any edition by Brown, LeMay and others.
**Grading Policy:**

Grades are based upon performance. See the table below.

<table>
<thead>
<tr>
<th>Assessment</th>
<th>Points</th>
<th>Due Date(s)</th>
<th>Points Each</th>
</tr>
</thead>
<tbody>
<tr>
<td>Exam 1-Review Material</td>
<td>100</td>
<td>9/18 4:45 PM</td>
<td></td>
</tr>
<tr>
<td>Exam 2</td>
<td>100</td>
<td>10/9 4:45 PM</td>
<td></td>
</tr>
<tr>
<td>Exam 3</td>
<td>100</td>
<td>10/31 4:45 PM</td>
<td></td>
</tr>
<tr>
<td>Exam 4</td>
<td>100</td>
<td>11/15 8:50 AM</td>
<td></td>
</tr>
<tr>
<td>Comprehensive Final Exam</td>
<td>100</td>
<td>Monday 12/11 8-10 AM</td>
<td></td>
</tr>
<tr>
<td>Laboratory Assignments</td>
<td>150</td>
<td>Due the next Monday at recitation</td>
<td>15 each</td>
</tr>
<tr>
<td>Formal Lab Report #1--Peer Review</td>
<td>25</td>
<td>10/12 or 10/13 by the end of laboratory</td>
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<tr>
<td>Formal Lab Report #1</td>
<td>50</td>
<td>Due 10/20 by 5pm</td>
<td></td>
</tr>
<tr>
<td>Formal Lab Report #2--Peer Review</td>
<td>25</td>
<td>11/16 or 11/17 by the end of laboratory</td>
<td></td>
</tr>
<tr>
<td>Formal Lab Report #2</td>
<td>50</td>
<td>Due 12/1 by 4pm</td>
<td></td>
</tr>
<tr>
<td>Electronic Homework</td>
<td>100</td>
<td>Before 12/7 at midnight</td>
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</tr>
<tr>
<td>On-line Final</td>
<td>100</td>
<td>Before 12/8 at midnight</td>
<td></td>
</tr>
</tbody>
</table>

*In class Exams*- See the course policies regarding the in-class exams. If a student misses an exam, the score on the final will replace their missed exam.

*On-line Exams*- See the course policies regarding the on-line exams. The on-line exams will be given via darrellrfry.com using Safe Exam Browser.

*Electronic Homework* – It is STRONGLY recommended that you keep pace with the assigned due dates if you desired to pass the course. All homework is due Any computer capable of connecting to the internet can assess the course at darrellrfry.com Enter the following to log into the system:

- **Select course:** CHEM 3421
- **username:** Use your SFA user name (username within SFA email address)
  
i.e. SFA email address: lastnamei@jacks.sfasu.edu    username: lastnameny
- **password:** $Sfa$ plus your student id number with no spaces
  
i.e. id# 12345678    password: $Sfa12345678$ (note $ sign and capital S)

*Formal Laboratory Reports*- Comprehensive rubrics for each formal lab report will be provided. Students who are late with the formal lab reports (peer review and/or the report) will be penalized 15 percent per day. After 5 days, the work will not be accepted.

*Final Exam*- The final exam will be comprehensive. The final exam may be a standardize exam.
Grading scale - A= 90 - 100%; B= 80 - 89%; C= 60 - 79%; D= 50 - 59%; F= below 50% -- please note, students must also have met the course requirements which are given below, in addition to having the percentages given here to earn the grades articulated.

Course Requirements

In order to receive a C or above for the course, in addition to earning the points required students must also meet the following requirements.

1. Students must attend at least 85% of the lecture time.
2. Students must complete at least 95% of the homework assignments.
3. Students must score at least a 70% on the homework assignments.
4. Students must take the comprehensive final.
5. Students must take at least three of the in-person exams. (Please note, students are strongly encouraged to take all of the in-person exams! In the event that they miss one exam due to a university approved excused absence, the final will replace the missed exam.)
6. All laboratory assignments must be completed. (Including calculations, discussion, etc…)
7. All worksheets given in the recitation must be completed AND without major errors or emissions.
<table>
<thead>
<tr>
<th>Week</th>
<th>Dates</th>
<th>PowerPoints</th>
<th>Work Sheets</th>
<th>Harris Chapters</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>8/28-9/1</td>
<td>1-ionic nomenclature</td>
<td>a-nomenclature &amp; REDOX</td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>2-REDOX with the ion electron method</td>
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<tr>
<td></td>
<td></td>
<td>3-Concentration Units</td>
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<td></td>
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<tr>
<td>2</td>
<td>9/4-9/8</td>
<td>4-Acids and Bases Review</td>
<td>b-concentration problems</td>
<td></td>
</tr>
<tr>
<td>3</td>
<td>9/11-9/15</td>
<td>5-Equilibrium</td>
<td>c-acid base and equilibrium</td>
<td></td>
</tr>
<tr>
<td>EXAM 1</td>
<td>9/18</td>
<td>1-5</td>
<td></td>
<td>a-c</td>
</tr>
<tr>
<td>4</td>
<td>9/18-9/22</td>
<td>6-The Analytical Process</td>
<td>d-correct Exam 1</td>
<td></td>
</tr>
<tr>
<td>5</td>
<td>9/25-9/29</td>
<td>7-Error</td>
<td>e-error</td>
<td></td>
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<tr>
<td>6</td>
<td>10/2-10/6</td>
<td>8-Statistics</td>
<td>f-statistics</td>
<td></td>
</tr>
<tr>
<td>EXAM 2</td>
<td>10/9</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>7</td>
<td>10/9-10/13</td>
<td>9-Titrations</td>
<td>g-Titrations</td>
<td></td>
</tr>
<tr>
<td>8</td>
<td>10/16-10/20</td>
<td>10-Diprotic Systems</td>
<td>h-Diprotic Systems</td>
<td></td>
</tr>
<tr>
<td>9</td>
<td>10/23-10/27</td>
<td>11-Systematic Treatment</td>
<td>i-Systematic Treatment</td>
<td></td>
</tr>
<tr>
<td>Exam 3</td>
<td>10/30</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>10</td>
<td>10/30-11/3</td>
<td>12-Spectroscopy</td>
<td>j-Spectroscopy 1</td>
<td></td>
</tr>
<tr>
<td>11</td>
<td>11/6-11/10</td>
<td></td>
<td>k-Spectroscopy 2</td>
<td></td>
</tr>
<tr>
<td>12</td>
<td>11/13-11/17</td>
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<tr>
<td>Exam 4</td>
<td>11/15</td>
<td></td>
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<tr>
<td>13</td>
<td>11/20-11/24</td>
<td>Thanksgiving Break</td>
<td></td>
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</tr>
<tr>
<td>14</td>
<td>11/27-12/1</td>
<td>13-Electrochemistry</td>
<td>l-Electrochemistry</td>
<td></td>
</tr>
<tr>
<td>15</td>
<td>12/4-12/8</td>
<td>14-Separations</td>
<td>m-Separations</td>
<td></td>
</tr>
<tr>
<td>16</td>
<td>12/11-12/15</td>
<td>Final Exam Week</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

Students are expected to fill out the far-right column on the table above.
<table>
<thead>
<tr>
<th>THURSDAY LABORATORY DATE</th>
<th>FRIDAY LABORATORY DATE</th>
<th>TASK</th>
<th>DUE DATE FOR BLUE BOOK</th>
</tr>
</thead>
<tbody>
<tr>
<td>8/31</td>
<td>9/1</td>
<td>1-Introduction to Microsoft Excel[^]</td>
<td>9/4</td>
</tr>
<tr>
<td>9/7</td>
<td>9/8</td>
<td>2-Check-in and Massing Pennies</td>
<td>9/11</td>
</tr>
<tr>
<td>9/14</td>
<td>9/15</td>
<td>3-Make and Standardize Sodium Hydroxide</td>
<td>9/18</td>
</tr>
<tr>
<td>9/21</td>
<td>9/22</td>
<td>3-continued</td>
<td>9/25</td>
</tr>
<tr>
<td>9/28</td>
<td>9/29</td>
<td>4-Determination of the molecular weight of an unknown monoprotic weak acid</td>
<td>10/2</td>
</tr>
<tr>
<td>10/5</td>
<td>10/6</td>
<td>5-Determination of the pKₐ of an unknown monoprotic weak acid</td>
<td>10/9</td>
</tr>
<tr>
<td>10/12</td>
<td>10/13</td>
<td>Peer Review of Formal Lab Report #1—due before the end of laboratory</td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>Formal Laboratory Report Due Friday 10/20 by 5 PM</td>
<td></td>
</tr>
<tr>
<td>10/19</td>
<td>10/20</td>
<td>Round Robin 6-9: Dry Lab, UV-VIS, Vernier Salt Solution, Vernier Food Dye.</td>
<td>10/23</td>
</tr>
<tr>
<td>10/26</td>
<td>10/27</td>
<td>Round Robin 6-9: Dry Lab, UV-VIS, Vernier Salt Solution, Vernier Food Dye.</td>
<td>10/30</td>
</tr>
<tr>
<td>11/2</td>
<td>11/3</td>
<td>Round Robin 6-9: Dry Lab, UV-VIS, Vernier Salt Solution, Vernier Food Dye.</td>
<td>11/6</td>
</tr>
<tr>
<td>11/16</td>
<td>11/17</td>
<td>Peer Review of Formal Lab Report #2—due before the end of laboratory</td>
<td></td>
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<tr>
<td></td>
<td></td>
<td>Formal Laboratory Report Due Friday 12/1 by 5 PM</td>
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<tr>
<td></td>
<td></td>
<td>Students can earn 10 extra points if they turn in their Formal Lab Report #2 by 11/24.</td>
<td></td>
</tr>
<tr>
<td>11/23</td>
<td>11/24</td>
<td>Thanksgiving Break-no laboratory</td>
<td></td>
</tr>
<tr>
<td>11/30</td>
<td>12/1</td>
<td>10-Check out</td>
<td>11/30 OR 12/1</td>
</tr>
<tr>
<td>12/7</td>
<td>12/8</td>
<td>On-line Lab Final</td>
<td></td>
</tr>
</tbody>
</table>

All experiments are done as an individual EXCEPT for the UV-VIS experiment which will be done in a group of no larger than 4. The groups will be assigned by Dr. Fry.

Labs 3, 4 and 5 will be combined into a formal lab report. The purpose of the three laboratories is to identify a weak monoprotic acid. In order to do so, students make and standardize their own sodium hydroxide in lab 3. For laboratories 4 and 5, students will be given a SINGLE UNKNOWN. Students are responsible for keeping up with their own unknown. (Store it in your drawer.) In lab 4, students use an indicator (phenolphthalein) to determine the molecular weight of their unknown monoprotic weak acid. In lab 5, students will use a pH meter to determine the pKₐ of their unknown monoprotic weak acid.

A formal written lab report will be made for the UV-VIS laboratory.

[^] The Introduction to Microsoft Excel is given asynchronously via video.
CLASS POLICIES:

1. If (or when) Dr. Fry makes a mistake, give him a chance to fix it. I own up to my mistakes and I’m quick to fix them. However, when a student goes around me, without giving the chance to fix it, then I’m not as quick fix the mistake. Once someone else is involved, I have to carefully keep them informed. This takes time! And this time is time that could have been better spent with you and me working out the problem. Again, if you have a problem with the course (or me), come talk with me—not someone else!

2. Knowing your grade, and how the grade will be calculated, is your responsibility!

3. Assigned seats will be given. Assigned seats may change during the semester. If you have a need or desire to sit in the front, please contact Dr. Fry at frydr@sfasu.edu and request a seat.

4. Attendance is required. Attendance is taken by student sitting in their assigned seats. If you are tardy, immediately after class, speak in person with Dr. Fry. Two tardy class days count as a single absence. Students with excessive absences will earn a F (or QF) for the course. Excessive absences are in excess of missing 85% of the classes.

5. Different assigned seats may be given for in-person exams. Moreover, Dr. Fry (or another proctor) is allowed to have a student move during the exam to another seat, without explaining why.

6. Students must complete at least 95% of the on-line homework. Moreover, students must score above a 70% on the on-line homework. Students who do not complete 95% and/or have less than a 70% on the on-line homework will be assigned a grade of F (or QF).

7. We will be following the attendance policy approved by the University. It can be found at the website below. https://www.sfasu.edu/ethehub/sos/notification-request
   In short, excused absences are only given for 4 reasons:
   1. Death of a family member*
   2. Admission to a hospital
   3. Personal emergency (car accident, drastic life event such as a fire)*
   4. Administrative and other (jury duty, court subpoena, etc.)
   *
   Other situations may be considered at the discretion of the dean of students.

   Students with excess absences may receive an F (or QF) for the course.

9. Files that are submitted to Dr. Fry must be saved as a pdf. For the file name use your last name and the initial of your first.

10. A paragraph is defined as a well written set of 6-8 complete and grammatically correct sentences that are related. When asked to produce a paragraph, please use the definition above.

11. Many of the problems are worked best through the factor label method. Take care to make your work neat and conform to the guidelines presented in class to receive credit (partial or full). Students not using the factor label method (as described in class) will not receive credit for problems worked with the factor label method.

12. I do not discuss student grade(s) for the course in person. Instead, all communication about grades will be via D2L (to communicate your score and give feedback) or through email. Please email me at frydr@sfasu.edu. Write out your question/complaint and explain why it may have been mis-graded. Include a digital copy of the entire assignment. Please note, the entire assignment will be regraded, so your grade may go up or down.

13. I do not discuss the schedule, instead I will use written communication such that everyone has equal access to the schedule. The tentative schedule is found in the syllabus. I will send email reminders as appropriate.

14. Keep all of your graded work and compare the score on your paper with that listed on D2L. If there is a discrepancy, bring it to my attention via email ASAP.

15. Do not email Dr. Fry through the D2L system. Instead send him email at frydr@sfasu.edu.

16. Official notices will be given through your jacks account and/or the D2L system.
   a. In the event of a cyber-attack, Dr. Fry will post information on darrellfrfy.com. Moreover, you can contact the Chemistry Department and leave Dr. Fry a message by calling 936 468 3606.
17. Students who know in advance that they will miss class, must make arrangements to get the exams (or assignments) well before they leave!
   a. This includes student athletes.
   b. This includes students who are attending a professional meeting.
18. Students who wish to use Disability Services must meet with Dr. Fry and discuss their needs. If the student must have exams proctored in Disability Services, then they will need to take the exam BEFORE the class does.
   Moreover, the exam they take may be a different version of what the other students will take.
19. **In-Person Exam Policies**—In-person (or in-class) exams will be closely monitored. Students who do not follow the rules will receive a zero for the exam (without a make-up).
   a. The proctor has control of the room. If they want a student to move, then the student should move. If they say: “Hey, we all need to leave”, then we all need to leave. We will figure out the exam later.
   b. No cheating!!
   c. All books, back packs, etc... must be placed at the front or back of the room. The student should only have their calculator (that cannot connect to the internet), a pencil (or pencils), an eraser.
   d. Scratch paper will be provided; a periodic table will be provided; a version of the solubility rules will be provided.
   e. Assigned seats may be given. Find your seat and sit in it.
20. **On-line Exam Policies**—I have limited control over what students actually do during the on-line exam. However, some policies are listed below.
   a. Work alone; students must not receive any outside help during the on-line exams. This includes students, proctors, faculty, AAR personnel, parents, siblings, etc...If you are caught using another person, you will receive a zero for the exam.
   b. Exams are time limited. You will have enough time to do the work; however, you will not have enough time to look up how to do a problem or a specific fact.
   c. Students must use only one computer (or phone) during the online exam. Moreover, the safe exam browser must be used during the exam. The safe exam browser blocks students from accessing any sites. The activity logs will be checked after the exam. If students logged in with two devices while taking the on-line exam, they will receive a grade of zero. Make sure to log off all devices before taking the exam!
   **d. With the policies above in mind, you may use your own printed text (not an electronic version) and your own hand written notes during the on-line exams.**
21. Students are expected to conduct themselves as responsible scholars while in class. If you are disrupting class, I will kick you out of class and do my best to make sure you do not disrupt my class again. Some additional guidelines are given below.
   a. Turn off your cell phones when you are in class.
   b. You are not allowed to read other materials (e.g. newspapers) or study for another course while in class.
   c. You are not allowed to sleep in class.
My personal thoughts are given below. Please read this, and think it through!

I’ve been teaching at the collegiate level for over twenty years. The first group of students I taught are now in their 40’s. Most of them have wonderful lives: children, marriages, homes, and successful careers! Twenty years is a long time to see how small things lead can lead to either wonderful outcomes or very challenging outcomes. As we grow older, our responsibilities increase: a spouse, children, careers, aging parents, etc. But, for most of you, right now is a very special time! It is a time when you can work on increasing your capacity. I encourage you to increase your capacity, so that when your responsibilities increase, you will be able to meet the challenge. In order to do this, you need to do the following.

1) Get a good night’s sleep! A good night’s sleep allows you to attend every class with your brain fully engaged on your course work. In order to be fully engaged (in any task) I suggest you turn off your cell phone!

2) Cultivate close relationships with people who have similar aspirations as yourself. If your aspirations differ from others, recognize this! And, most importantly, keep people with lower aspirations than yourself at arm’s length. They will be a distraction to you reaching your goals. Conversely, those with similar aspirations as yourself, should be embraced! They will help you meet your goals!

3) What you do outside of class matters! The human race knows how to efficiently learn things. We know that we cannot pay attention for long periods of time. We know that stories motivate and communicate a paradigm efficiently. You can reinvent the wheel if you like, but I’d do the following first:
   i. Small bouts of challenges are very efficient at learning. Use notes cards to do this!
   ii. Learn to read (start with the textbook!). Reading has proven to be the most efficient way of communicating nuisances. The future will be full of nuisances. Watching a video, or engaging in a lecture, only takes you so far. Reading is the way you really learn.
   iii. Rewriting your notes, with the text book in front of you, before the next class period is a very, very efficient way to learn something!
   iv. Start working the homework as soon as possible. If you get stuck on a type of problem, you can come by my office where I can help. The most efficient person at helping you for this course is Dr. Fry!! Use his office hours! Email him with questions!
   v. Learn to write better! Writing has proven to be the most efficient way for a person to communicate their ideas. In the future, you will greatly desire for others to listen to your ideas! And the proven way of doing this is through writing! (Not social media, not videos, etc…) Writing takes time. Writing takes practice.

ATTENDANCE POLICY:

Students must be completing homework assignments by the due date to be considered actively participating in the class. Attendance is required for all exam dates and will be given during the assigned times unless other arrangements are approved by the instructor prior to the scheduled exam day. There are no make-ups exams for notifications given the day of the exam. Students registered for face-to-face modality sections are required to attend all lecture sessions. Students who do not attend at least 85% of the lecture time will earn an F (or QF) for the course.
ACADEMIC INTEGRITY (4.1):

Academic integrity is a responsibility of all university faculty and students. Faculty members promote academic integrity in multiple ways including instruction on the components of academic honesty, as well as abiding by university policy on penalties for cheating and plagiarism.

The Code of Student Conduct and Academic Integrity outlines the prohibited conduct by any student enrolled in a course at SFA. It is the responsibility of all members of all faculty, staff, and students to adhere to and uphold this policy.

Articles IV, VI, and VII of the new Code of Student Conduct and Academic Integrity outline the violations and procedures concerning academic conduct, including cheating, plagiarism, collusion, and misrepresentation. Cheating includes, but is not limited to: (1) Copying from the test paper (or other assignment) of another student, (2) Possession and/or use during a test of materials that are not authorized by the person giving the test, (3) Using, obtaining, or attempting to obtain by any means the whole or any part of a non-administered test, test key, homework solution, or computer program, or using a test that has been administered in prior classes or semesters without permission of the Faculty member, (4) Substituting for another person, or permitting another person to substitute for one’s self, to take a test, (5) Falsifying research data, laboratory reports, and/or other records or academic work offered for credit, (6) Using any sort of unauthorized resources or technology in completion of educational activities.

Plagiarism is the appropriation of material that is attributable in whole or in part to another source or the use of one’s own previous work in another context without citing that it was used previously, without any indication of the original source, including words, ideas, illustrations, structure, computer code, and other expression or media, and presenting that material as one’s own academic work being offered for credit or in conjunction with a program course or degree requirements.

Collusion is the unauthorized collaboration with another person in preparing academic assignments offered for credit or collaboration with another person to commit a violation of any provision of the rules on academic dishonesty, including disclosing and/or distributing the contents of an exam.

Misrepresentation is providing false grades or résumés; providing false or misleading information in an effort to receive a postponement or an extension on a test, quiz, or other assignment for the purpose of obtaining an academic or financial benefit for oneself or another individual or to injure another student academically or financially.

Examples of academic dishonesty include:

- Fabricating measurements or data in the laboratory
- Copying or lifting or otherwise not writing their own lab reports (all or part)
- Exchanging answers or information during a test or quiz
- Looking at another student’s paper during a test or quiz
- Bringing or looking at a book or other unauthorized source during the quiz or test
- If caught cheating a zero will be given to the cheater and or both students colluding in cheating.

Students engaging in any type of academic misconduct (including, but not limited to: cheating, plagiarism, or any other action that can improperly affect my evaluation of your performance) will be
subject to sanctions in accordance with SFA Academic Integrity Policies. Please note: The usage of electronic devices (including, but not limited to: cell phones, PDAs, mp3 players, etc.) while a quiz or exam is being given will be treated as academic misconduct. DO NOT HAVE THESE DEVICES OUT DURING A QUIZ OR AN EXAM! I will recommend a grade of "F" for the course.

WITHHELD GRADES SEMESTER GRADES POLICY (5.5):

Ordinarily, at the discretion of the instructor of record and with the approval of the academic chair/director, a grade of WH will be assigned only if the student cannot complete the coursework because of unavoidable circumstances. Students must complete the work within one calendar year from the end of the semester in which they receive a WH, or the grade automatically becomes an F. If students register for the same course in future terms the WH will automatically become an F and will be counted as a repeated course to compute the grade point average. For additional information, go to https://www.sfasu.edu/policies/course-grades-5.5.pdf.

STUDENTS WITH DISABILITIES:

To obtain disability related accommodations, alternate formats and/or auxiliary aids, students with disabilities must contact the Office of Disability Services (ODS), Human Services Building, and Room 325, 468-3004 / 468-1004 (TDD) as early as possible in the semester. Once verified, ODS will notify the course instructor and outline the accommodation and/or auxiliary aids to be provided. Failure to request services in a timely manner may delay your accommodations. For additional information, go to http://www.sfasu.edu/disabilityservices/.

Student Wellness and Well-Being

SFA values students’ overall well-being, mental health and the role it plays in academic and overall student success. Students may experience stressors that can impact both their academic experience and their personal well-being. These may include academic pressure and challenges associated with relationships, emotional well-being, alcohol and other drugs, identities, finances, etc.

If you are experiencing concerns, seeking help, SFA provides a variety of resources to support students’ mental health and wellness. Many of these resources are free, and all of them are confidential.

On-campus Resources:

The Dean of Students Office (Rusk Building, 3rd floor lobby)
www.sfasu.edu/deanofstudents
936.468.7249
dos@sfasu.edu

SFA Human Services Counseling Clinic Human Services, Room 202
www.sfasu.edu/humanservices/139.asp
936.468.1041

The Health and Wellness Hub “The Hub”
Location: corner of E. College and Raguet St.

To support the health and well-being of every Lumberjack, the Health and Wellness Hub offers comprehensive services that treat the whole person – mind, body and spirit. Services include:
• Health Services
• Counseling Services
• Student Outreach and Support
• Food Pantry
• Wellness Coaching
• Alcohol and Other Drug Education

www.sfasu.edu/thehub
936.468.4008
thehub@sfasu.edu

Crisis Resources:
• Burke 24-hour crisis line: 1.800.392.8343
• National Suicide Crisis Prevention: 9-8-8
• Suicide Prevention Lifeline: 1.800.273.TALK (8255)
• johCrisis Text Line: Text HELLO to 741-741

The instructor reserves the right to modify any part of this syllabus as circumstances dictate.
Changes will be communicated through e-mail.
Darrell R. Fry 8/21/2023